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HEALTH EFFECTS DIVISION SCIENTIFIC DATA REVIEWS EPA SERIES 361

"INERT INGREDIENTS OF PESTICIDE FORMULATIONS"
(FORMAT FOR IDENTIFICATION AND TOXICOLOGICAL APPRAISAL)

Format

A. EPA Accession Number and Name:

000263 Stearic acid, calcium salt of

B. American Chemical Society Chemical Abstracts Service (CAS) Name and Registry Number

Octadecanoic acid, calcium salt 1592-23-0

- C. Other Names
 - (EPA S): Calcium stearate (S): Calcium distearate
 - (T): Aquacal; Calstar; Flexichem CS; Synpro stearate
- D. Chemical Composition

Ca(C18-H35-O2)2 (MW) 607.00

- E. EPA Chemical Code
- F. Molecular Structure (C17 H35 CO O)2 Ca
- G. Chemical and Physical Properties (1)
 - solubility practically insoluble in water, ether, chloroform, acetone, cold alcohol; slightly soluble in hot alcohol, hot vegetable and mineral oils
 - 2. specific gravity (or density) bulk density = 20 lb/ft3 (approx.)
 - 3. state, color, odor, etc. granular, fatty powder
 - 4. MP, BP, VP = (MP) $147^{\circ}-149^{\circ}$
 - corrosiveness
 - 6. technical products & impurities Calcium stearate is the calcium salt of a mixture of fatty acids in which stearic acid predominates.
 - 7. stability

- H. Use as an Inert solid diluent; carrier (3)
- I. Other Uses Active? Yes () No (X)
 for waterproofing fabrics, cement, stucco; explosives; stablizer in PVC resins; as a lubricant; pencils and wax crayons (1)
 (Pharmaceutical) lubricant
- J. Government Regulations May be safely used in foods as a defoaming agent subject to FDA 21 CFR 121.1099. (4) May be safely used in foods subject to FDA 21 CFR 121.1071. (5)

 EPA 40 CFR 180.1001 Exempt from the requirement of a tolerance when used in accordance with good agricultural practice as an inert ingredient in pesticide formulations applied to growing crops or raw agricultural products after harvest. (3)
- K. <u>Manufacturer(s)</u> Conray Products; Ferro Corp.; C.P. Hall; Mallinokrodt; and several other manufacturers and suppliers (6)
- L. Environment
- M. Toxicology "...stearic acid and its salts are normal products of the metabolism of fats and their metabolic fate is well established. Provided the contribution of the cation does not add excessively to the normal body load, there is no need to consider the use of these substances in any different light to that of dietary fatty acids." (7)
- N. Recommendation: Class 4

 Excessive feeding levels of stearates have resulted in fatty deposits in the liver without histopathological evidence of damage. Such deposits disappear upon reduction of the feeding level. Other than problems associated with the ingestion of high levels of saturated fats, no adverse biological activity is expected.
- O. Sources Used In Search:
 - 1. On Line Data Bases
 - a. Toxline
 - b. Medline
 - c. Chemline

- 2. Major References
 - a. The Merck Index
 - b. NIOSH Registry of Toxic Substances
 - c. Chemical Abstracts
 - d. Biological Abstracts

P. References & Review Articles

- (1) Merck and Co., The Merck Index, Rahway, N.J.: 1976.
- (2) National Institute for Occupational Safety and Health, Registry of Toxic Effects of Chemical Substances, Washington, D.C.: 1976.
- (3) U.S. E.P.A., Code of Federal Regulations; 40, part 180.1001, Washington, D.C.: 1976.
- (4) U.S. F.D.A., <u>Code of Federal Regulations</u>; <u>21</u>, part 121.1099, Washington, D.C.: 1976.
- (5) Ibid, part 121.1071.
- (6) Oil, Paint and Drug Reporter, OPD Chemical Buyers Directory, Schnell Publishing Co., New York: 1974.
- (7) World Health Organization, Toxicological evaluation of some food additives including anticaking agents, antimicrobials, autioxidants emulsifiers and thickening agents, WHO Food Additives Series; 5, pp 20, Geneva: 1974.



R100910

Chemical:

Calcium stearate

PC Code:

800119

HED File Code

13100 Other Tox Documents

Memo Date:

01/01/87

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